

Cell Processes

Chapter 3

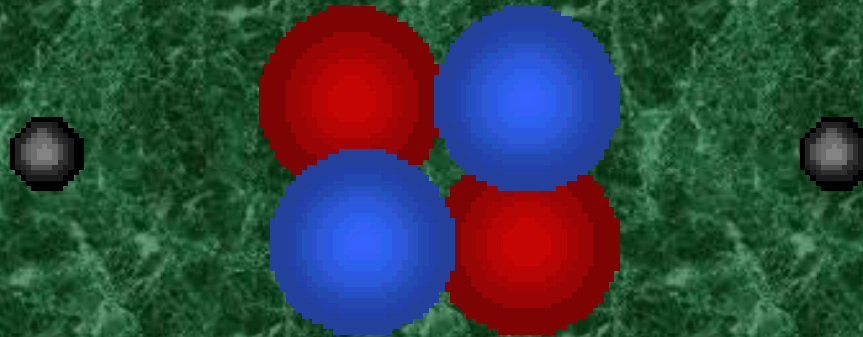
Chemistry of living things

- Living things are made of matter
- Matter is any thing that has mass and takes up space

What matter is made of

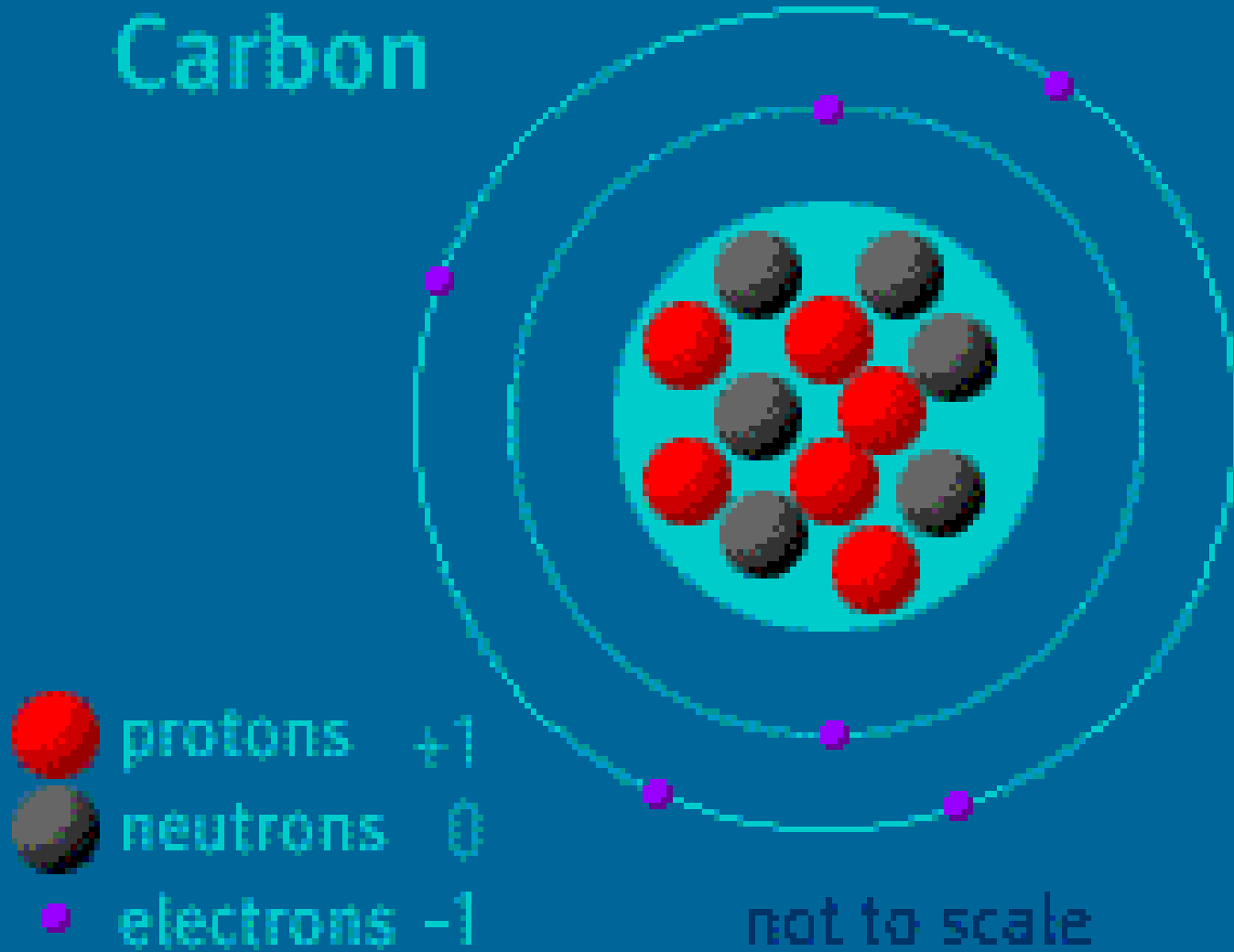
- **Atom is the smallest unit of matter**
- **Center nucleus of the atom**
 - **Proton - particle that has a positive charge**
 - **Neutron – particle that has no charge**
 - **Electron cloud – area around the nucleus**
 - **Electron – negatively charged particle that moves around the nucleus at the speed of light**

Model of the Atom



Model of Atom

Carbon



Atomic Particle Table

	Mass	Charge
Protons	1 AMU	Positive
Neutrons	1 AMU	None
Electrons	0 AMU	Negative

- **Energy is released by atoms being rearranged or broken apart in a chemical reaction.**

- **Element – only one type of atom present in a substance**

- **Elements can't be broken down any further by simple chemical processes**

- **Each element has its own symbol**

Compounds – is the result of atoms combining

- Atoms combine in two ways
 - Atoms share their outer most electrons in a covalent bond
 - Atoms become electrically charged because one loses an electron and another one gains an electron and so they combine to form an ionic bond.

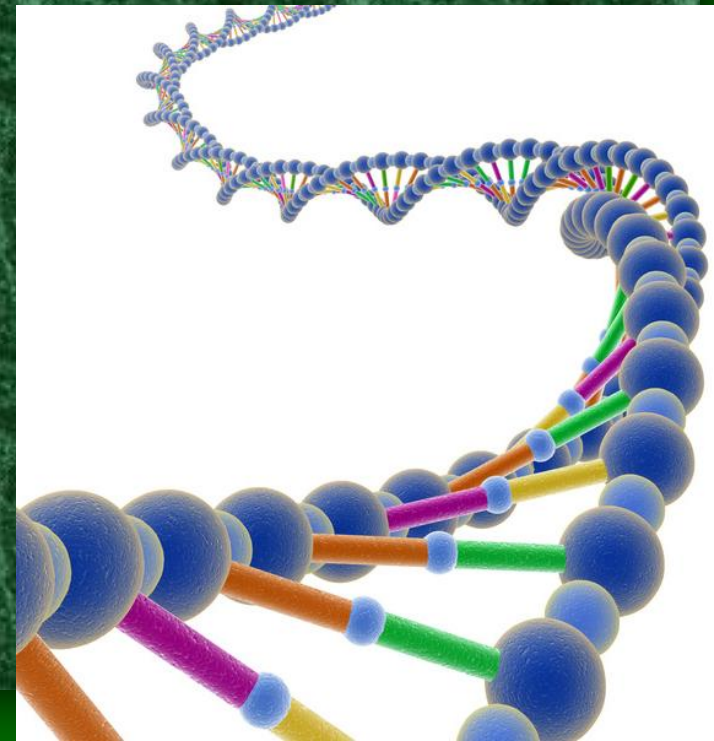
Mixtures –

combination of substances in which the individual substances retain their own properties

- **Solution – a mixture in which one or more substances mix evenly with other substances. Particles do not settle out (salt water)**
- **Suspension – A mixture in which substances are evenly spread throughout a liquid or gas. Particles in a suspension settle out over time (paint)**

Compounds in living organisms

- Organic – contain carbon
- Inorganic – not ever living
- Organic compounds that make up all living things are
 - Carbohydrates
 - Sugars and starches
 - Lipids
 - Fats, oils and waxes
 - Proteins
 - Made up of amino acids
 - Building blocks
 - Nucleic acids
 - DNA
 - RNA



Inorganic Compounds – made of elements other than carbon

- 1. Water**
- 2. Carbon Dioxide**
- 3. Oxygen**

Water



- Living things are composed of more than 50% H_2O
- Water is polar
- Polar means that it has a positive end and a negative end
- This is why water washes things off you and helps move things through us
- This is why ice makes 6 sided snow flakes and make it float on liquid H_2O

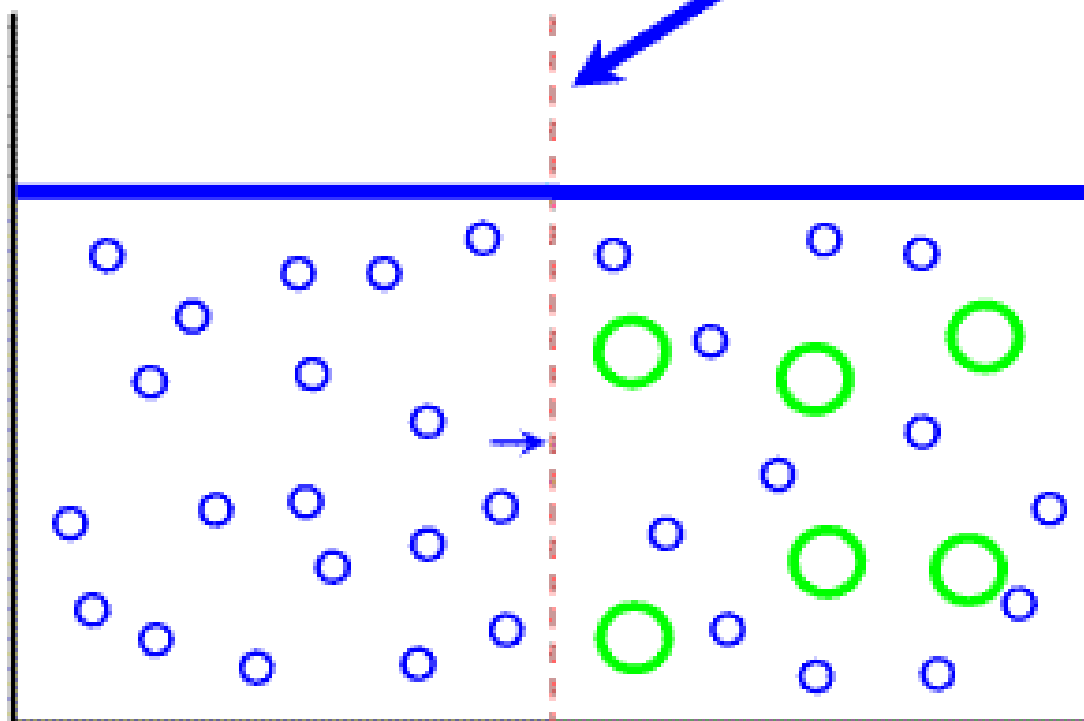
Cell Membrane

- allows certain material to pass through it while preventing others. This is called selective permeability.
- Cells obtain food, oxygen, etc. from their environment and release waste into the environment
- Small molecules and ions move randomly across the membrane from high to low concentration.
 - This is called diffusion – moving from high to low
 - Equilibrium – when the molecules of a substance are spread evenly throughout a space
 - Osmosis – is the diffusion of water.

Osmosis

○ - Water
○ - Sugar

Selectively Permeable Membrane



Low Sugar Concentration

High Water Concentration

High Sugar Concentration

Low Water Concentration

Transport – is the movement of substances across the cell membrane

Passive transport

- is a transport that requires no energy from the cell to move materials
 - Example: osmosis and diffusion
- Facilitated diffusion – diffusion where the transport protein molecules in cell membrane help in the transport of materials across the cell membrane

Active transport

- When energy is required to move through the cell membrane. Transport proteins are used here also.
 - This is when substances are moved from low concentration to high concentration.
 - The cells energy is required for this process
 - Example: plant roots taking minerals from the soil

Endocytosis

- **capture things too big to pass through membrane, surround the particle and form a vacuole**
 - **Phagocytosis- large particles moved into the cell**
 - **Pinocytosis – fluids moved into the cell**
 - **Exocytosis – opposite of endocytosis, cells move waste vacuoles or protein packages from golgi bodies out of the cell. This can also so be for waste removal.**

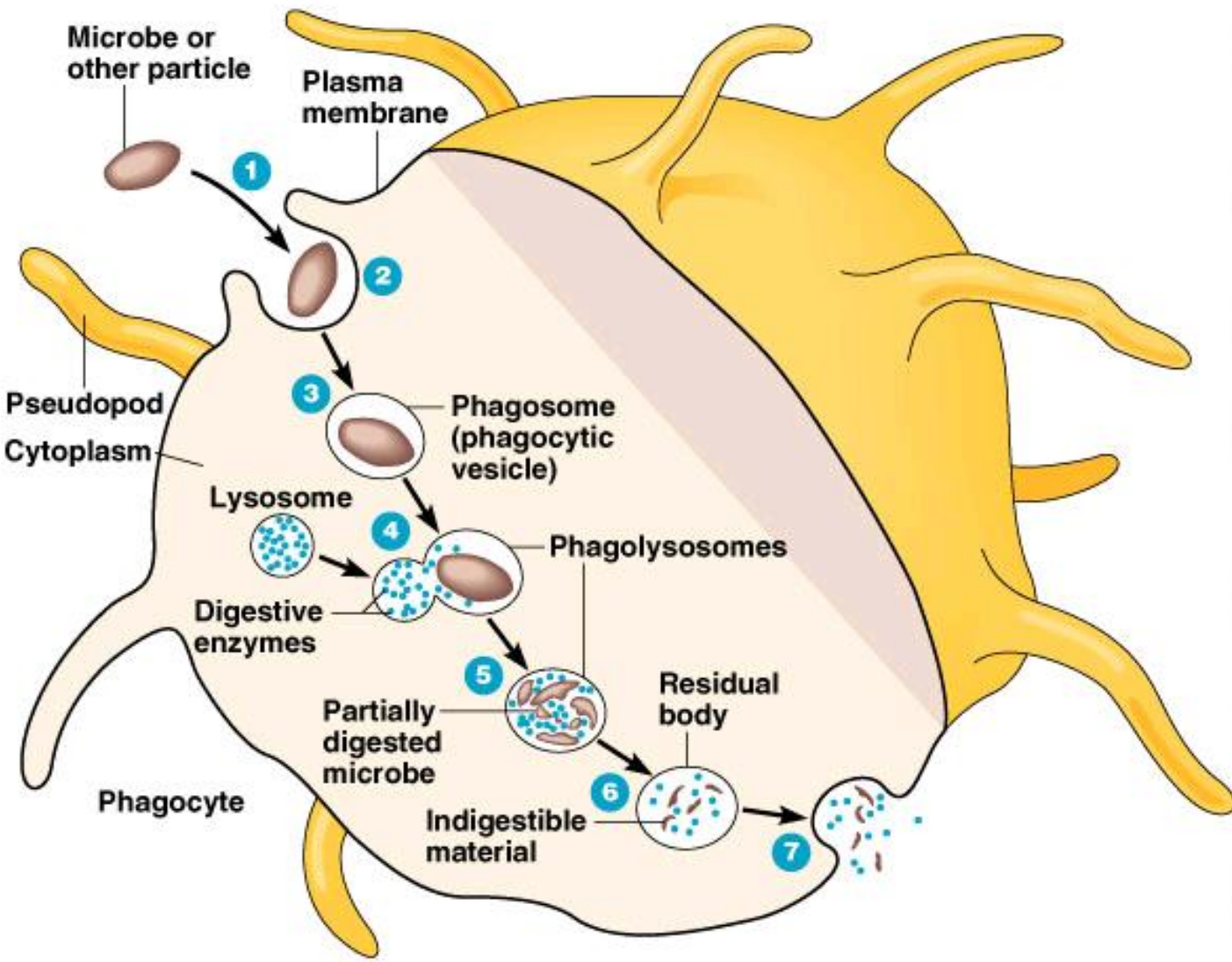
Carbohydrate chain

Lectin



Phagocyte

Bacteria



- 1** Chemotaxis and adherence of microbe to phagocyte.
- 2** Ingestion of microbe by phagocyte.
- 3** Formation of a phagosome.
- 4** Fusion of the phagosome with a lysosome to form a phagolysosome.
- 5** Digestion of ingested microbe by enzymes.
- 6** Formation of residual body containing indigestible material.
- 7** Discharge of waste materials.

(a) Phases of phagocytosis

Copyright © 2004 Pearson Education, Inc., publishing as Benjamin Cummings.

Metabolism – total of all chemical activities of an organism (to stay alive, grow, reproduce)

Producers

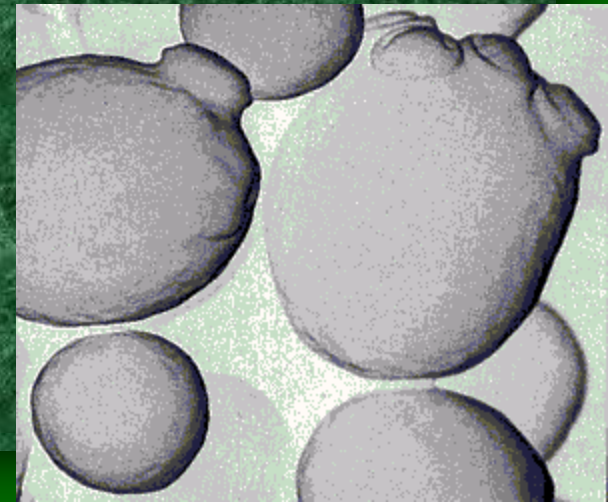
- **make their own food in the chloroplasts by a process called photosynthesis**
- **Photosynthesis – changes light energy into chemical energy**
 - $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
 - **carbon dioxide + water and sun's energy yield sugar + oxygen**
 - **the sun's energy is stored in the sugar molecule**

Consumers

- require another organism to assemble their food for them
- Cellular Respiration releases the energy from the sugar molecule to do cell processes
 - $\text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
 - Respiration is the opposite of photosynthesis
 - Respiration takes place in the mitochondria.

Fermentation

- A form of respiration that produces energy when oxygen is insufficient
 - Does not produce as much energy as regular respiration
 - Lactic acid fermentation Produces a by product – lactic acid
 - Alcoholic fermentation produces a by product – alcohol



Quiz

- 1. Write out the chemical equation for photosynthesis
- 2. Write out the chemical equation for respiration.
- 3. What is respiration without oxygen called?
- 4. What are two types of endocytosis?
- 5. What is an element?
- 6. What are the four organic compounds that make up living things?